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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : C08L 43/04, C08J 3/22	A1	(11) International Publication Number: WO 97/12936 (43) International Publication Date: 10 April 1997 (10.04.97)
(21) International Application Number: PCT/JP96/02712 (22) International Filing Date: 20 September 1996 (20.09.96) (30) Priority Data: 7/252123 29 September 1995 (29.09.95) JP (71) Applicant (for all designated States except US): SUMITOMO BAKELITE COMPANY LIMITED [JP/JP]; 5-8, Higashishinagawa 2-chome, Shinagawa-ku, Tokyo 140 (JP). (72) Inventor; and (75) Inventor/Applicant (for US only): MORI, Tsuneharu [JP/JP]; Sumitomowakakusaso 303, 5-5, Wakakusa 1-chome, Nishinomiya-shi, Hyogo 663 (JP). (74) Agents: ASAMURA, Kiyoshi et al.; Room 331, New Ohtemachi Building, 2-1, Ohtemachi 2-chome, Chiyoda-ku, Tokyo 100 (JP).		(81) Designated States: CA, CN, KR, SG, US, European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE). Published <i>With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
(54) Title: PROCESS FOR PRODUCING SILANE-CROSSLINKED POLYOLEFIN (57) Abstract A process for producing a silane-crosslinked polyolefin which is characterized by melt-blending (i) a polyolefinic base polymer, (ii) a carrier polymer A containing an organic unsaturated silane and a free radical generating agent and containing substantially no water and (iii) a carrier polymer B containing a silanol condensation catalyst and an antioxidant, at a temperature higher than the crystal melting point of the base polymer to effect reaction, and then contacting the resulting reaction product with water to effect crosslinking. The process can provide a silane-crosslinked polyolefin excellent in extrusion processability as well as in crosslinking property, mechanical property and heat resistance.		

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DESCRIPTION

PROCESS FOR PRODUCING SILANE-CROSSLINKED
POLYOLEFIN

Technical Field

The present invention relates, in the silane-crosslinking of polyolefins, a process of silane-crosslinking for producing a silane-crosslinked
5 polyolefin in one step which uses a carrier polymer A containing an organic unsaturated silane and the like in a high concentration and a carrier polymer B containing a silanol condensation catalyst and the like.

10 Background Art

A simple method which has been widely known for crosslinking polyolefins is the so-called silane crosslinking method which comprises grafting an organic unsaturated silane to the polyolefin in the presence of
15 a free radical generating agent to effect silane grafting, and then contacting the resulting silane-grafted polymer with water in the presence of a silanol condensation catalyst to effect crosslinking. This method is disclosed, for example, in JP-B-48-1711 and
20 JP-A-57-49109.

However, this method comprises at least two steps, that is, the silane grafting step and the silanol condensation step. Accordingly, at least two extrusion steps are necessary to obtain the ultimate
5 product, inevitably resulting in economical disadvantage.

An already known one-step process is the monosil process. This process requires a liquid addition apparatus for injecting an organic unsaturated
10 silane in the form of liquid into an extruder, and hence involves the problems of slippage and metering error. Moreover, the process requires an expensive special type extruder with a high L/D ratio to insure uniform dispersion of small amounts of additives,
15 resulting in economical disadvantage. Furthermore, a very high grade technique is necessary in the extrusion.

Another known one-step process is a silane crosslinking method which introduces silane into a
20 solid carrier polymer disclosed in JP-A-3-167229. In this method, however, a porous polymer or EVA is used as the solid carrier polymer and, in addition to silane and a free radical generating agent, such other additives as a silanol condensation catalyst and an
25 antioxidant are introduced to the solid carrier polymer; hence the method involves the problems of low crosslinking efficiency and poor storability caused by oligomerization by condensation of silane or inhibition

of crosslinking due to radical capture.

The present invention has been made to overcome the aforesaid problems and has for its object to provide, in the silane-crosslinking of polyolefins, a process of silane-crosslinking for producing a silane-crosslinked polyolefin in one step which uses a carrier polymer A containing an organic unsaturated silane and the like in high concentrations and a carrier polymer B containing a silanol condensation catalyst and the like.

Disclosure of the Invention

The present invention relates to a process for producing a silane-crosslinked polyolefin which is characterized by melt-blending (i) a polyolefinic base polymer, (ii) a carrier polymer A containing an organic unsaturated silane represented by the formula $RR'SiY_2$, wherein R is a monovalent olefinic unsaturated hydrocarbon group, Y is a hydrolyzable organic group, and R' is a monovalent hydrocarbon group other than aliphatic unsaturated hydrocarbon groups or is the same group as Y, and a free radical generating agent and containing substantially no water and (iii) a carrier polymer B containing a silanol condensation catalyst and an antioxidant, at a temperature higher than the crystal melting point of the base polymer to effect reaction, and then contacting the resulting reaction product with water to effect crosslinking. In this process for

producing a silane-crosslinked polyolefin, the base polymer is preferably a polymer selected from the group consisting of polyethylene, polypropylene, copolymer of ethylene and an α -olefin, ethylene-ethyl acrylate

5 copolymer (EEA), ethylene-methyl methacrylate copolymer (EMMA), ethylene-vinyl acetate copolymer (EVA), chlorinated polyethylene (CPE), and the mixtures thereof; the carrier polymer A is preferably a polymer selected from the group consisting of ethylene-ethyl

10 acrylate copolymer (EEA), ethylene-methyl methacrylate copolymer (EMMA), a hydrogenated block copolymer obtained by hydrogenating a block copolymer consisting essentially of a polymer block comprising at least one vinylaromatic compound as the principal monomer unit

15 and a polymer block comprising at least one conjugated diene compound as the principal monomer unit, and the mixtures thereof; the carrier polymer B is preferably a polymer selected from the group consisting of poly-

20 α -olefin, and the mixtures thereof; and the total amount of the carrier polymer A and the carrier polymer B is preferably 3-15% by weight.

Best Mode for Carrying Out the Invention

The present invention is described in detail

25 below.

The polyolefinic base polymer used in the present invention is not particularly limited and may

be, for example, common polyethylene, polypropylene, copolymer of ethylene and an α -olefin (wherein the α -olefin may be C₃-C₁₂ α -olefins, e.g., propylene, butene-1, pentene-1, octene-1, 4-methylpentene-1, 4-methylhexene-1, 4,4-dimethylpentene-1, nonene-1, decene-1, undecene-1 and dodecene-1), ethylene-ethyl acrylate copolymer (EEA), ethylene-methyl methacrylate copolymer (EMMA), ethylene-vinyl acetate copolymer (EVA), chlorinated polyethylene (CPE), and the mixtures thereof.

The organic unsaturated silane used in the present invention is grafted to the base resin to serve as the point which crosslinks the base resin molecules with each other. The organic unsaturated silane used in the present invention is a compound represented by the formula $RR'SiY_2$, wherein R is a monovalent olefinic unsaturated hydrocarbon group, Y is a hydrolyzable organic group, R' is a monovalent hydrocarbon group other than aliphatic unsaturated hydrocarbon groups or is the same group as Y. Preferably used is an organic unsaturated silane in which R' and Y are the same and which is represented by the formula $RSiY_3$, for example, vinyltrimethoxysilane, vinyltriethoxysilane, vinyltributoxysilane, allyltrimethoxysilane and allyltriethoxysilane.

The amount of the organic unsaturated silane to be added is 0.1-5% by weight, preferably 0.7-3% by weight, based on the total weight of polymers. When

the amount is less than 0.1% by weight, a sufficient grafting does not take place. An amount larger than 5% by weight may cause defective molding and also is economically disadvantageous.

- 5 The free radical generating agent used in the present invention acts as the initiator of silane grafting. The free radical generating agent used in the present invention may be various organic peroxides and peresters which have a strong polymerization
- 10 initiating effect, for example, dicumyl peroxide, α,α' -bis(t-butylperoxydiisopropyl)benzene, di-t-butyl peroxide, t-butyl cumyl peroxide, dibenzoyl peroxide, 2,5-dimethyl-2,5-bis(t-butylperoxy)hexane, t-butyl peroxy-pivalate, and t-butyl peroxy-2-ethylhexanoate.
- 15 The amount of the agent to be added is 0.01-0.5% by weight, preferably 0.05-0.2% by weight, relative to the total weight of polymers. When the amount is less than 0.01% by weight, the silane grafting does not proceed sufficiently. When the amount is larger than 0.5% by
- 20 weight, both the extrusion processability and the surface appearance of moldings tend to be poor.

 The free radical generating agent and the organic unsaturated silane can be incorporated into the carrier polymer A of the present invention by swelling

25 the carrier polymer A with a liquid mixture obtained by dissolving the free radical generating agent in the organic unsaturated silane. For incorporating the silane into polymer to a high concentration at this

time, the carrier polymer A needs to be preheated, but the temperature must not be higher than the crystal melting point of the polymer lest the polymer should undergo melting.

5 The carrier polymer A must be in the form of granules and must be compatible with the base polymer to be crosslinked and with the silane. The term "compatible" herein means that the carrier polymer A should not readily react with the silane and should be
10 dispersible or soluble in the base polymer. The suitable carrier polymer A is non-hygroscopic. Thus, the moisture absorption rate of the polymer is preferably relatively slow in order to minimize the possibility of premature hydrolysis and condensation of
15 the silane. In any way, substantially no water should be present in the carrier polymer A. The carrier polymer A used in the present invention is usually made into the form of granules, including pellets. Preferred is the form of pellets.

20 The carrier polymer A used in the present invention may be, for example, ethylene-ethyl acrylate copolymer (EEA), ethylene-methyl methacrylate copolymer (EMMA), a hydrogenated block copolymer obtained by hydrogenating a block copolymer consisting essentially
25 of a polymer block comprising at least one vinyl-aromatic compound as the principal monomer unit and a polymer block comprising at least one conjugated diene compound as the principal monomer unit, e.g.,

hydrogenated styrene-isoprene block copolymer (SEPS) and hydrogenated styrene-butadiene block copolymer (SEBS), and the mixtures thereof. Preferred of these is EEA or EMMA.

5 The silanol condensation catalyst, anti-oxidant, etc. may be incorporated into the carrier polymer B of the present invention by kneading them with the carrier polymer B, followed by granulation. The carrier polymer B should be in the form of granules
10 and should be a solid compatible with the base polymer to be crosslinked. The carrier polymer B is usually made into the form of granules, including pellets. Preferred is the form of pellets.

 The carrier polymer B has a softening point
15 equal to or higher than that of the carrier polymer A.

 The carrier polymer B used in the present invention may be, for example, polyethylene, polypropylene, copolymer of ethylene and an α -olefin, wherein the α -olefin may be C_3 - C_{12} α -olefins, e.g.,
20 propylene, butene-1, pentene-1, octene-1, 4-methylpentene-1, 4-methylhexene-1, 4,4-dimethylpentene-1, nonene-1, decene-1, undecene-1, dodecene, and the mixtures thereof.

 The silanol condensation catalyst used in the
25 present invention may be such organometallic compounds as dibutyltin dilaurate, stannous acetate, dibutyltin diacetate, dibutyltin dioctoate, lead naphthenate, zinc caprylate, cobalt naphthenate, tetrabutyl titanate,

lead stearate, zinc stearate, cadmium stearate, barium stearate and calcium stearate.

The amount of the silanol condensation catalyst to be added is 0.01-0.2% by weight, preferably 0.02-0.1% by weight, based on the total weight of polymers. When the amount is less than 0.01% by weight, the crosslinking reaction does not proceed sufficiently. When the amount is larger than 0.2% by weight, local crosslinking proceeds in the extruder at the time of extrusion, resulting in a greatly deteriorated appearance of the product. The silanol condensation catalyst should be incorporated into the carrier polymer B. This is because if the catalyst is incorporated into the carrier polymer A, oligomerization by condensation of the silane is promoted to cause deterioration of the appearance of products.

The antioxidant used in the present invention may be those conventionally used in processing polyolefins and is not particularly limited, but it should be incorporated into the carrier polymer B. This is because if it is incorporated into the carrier polymer A, crosslinking is inhibited owing to radical capture. Also when other additives are added, those additives which have the possibility of inhibiting crosslinking should be incorporated into the carrier polymer B.

The carrier polymers are added in such amounts that the sum of the amounts of the carrier polymers A and B is in the range of 3-15% by weight.

When it is less than 3% by weight, a sufficient grafting does not take place. An amount larger than 15% causes defective molding and is at the same time economically disadvantageous.

5 As other additives, if necessary and desired, conventionally used additives, for example, neutralizing agents, ultraviolet absorbers, antistatic agents, pigments, dispersants, thickeners, corrosion inhibitors, mildewproofing agents, flow regulators, other
10 inorganic fillers and other synthetic resins, may be added.

The present invention is described below with reference to Examples.

Preparation of carrier polymer A

15 According to the compounding ratio shown in Table 1, first a carrier polymer A was poured into a Super Mixer, mixed with stirring and preheated to 80°C. Then a liquid mixture obtained by dissolving a free radical generating agent in an unsaturated silane was
20 poured into the Super Mixer and, while stirring, the carrier polymer A was impregnated with the liquid mixture for 10 minutes.

Preparation of carrier polymer B

25 According to the compounding ratio shown in Table 2, a carrier polymer B, silanol condensation catalyst, antioxidant, etc. were kneaded by using a

pressure kneader and granulated.

* Materials used

- (1) EEA: ethylene-ethyl acrylate copolymer (EA content: 23% by weight)
- 5 (2) SEPS: hydrogenated styrene-isoprene block copolymer (styrene content: 30% by weight)
- (3) L-LDPE: linear low density polyethylene (density: 0.924 g/cm³, MI: 3.0 g/10 min)
- (4) VT MOS: vinyltrimethoxysilane
- 10 (5) DCP: dicumyl peroxide
- (6) LDPE: low density polyethylene (density: 0.925 g/cm³, MI: 1.5 g/10 min)
- (7) PP: polypropylene (homopolymer, MI(230°C): 2.0 g/10 min)
- 15 (8) DBTDL: Dibutyltin dilaurate
- (9) Antioxidant: phenolic antioxidant/Irganox 1010 (mfd. by Ciba Geigy Corp.)
- (10) Lubricant: low molecular weight polyethylene/Sanwax 171P (mfd. by Sanyo Chemical Industries, Ltd.)
- 20 (11) MDPE: medium density polyethylene (density: 0.930 g/cm³, MI: 2.0 g/10 min)

* Method of evaluation

- (12) Silane impregnability:

25 Impregnability was examined by heating and stirring the objective polymer with a VT MOS/DCP liquid mixture in a Super Mixer.

○ : well impregnable, × : not impregnable

(13) Extruded tape appearance

50 mm ϕ extruder, 120-150-170-180-170°C

L/D: 20, compression ratio: 3.5

tape die: width 100 m, lip gap 1 mm

- 5 Evaluation: Results were evaluated in the order of $\circ > \Delta > \times$, the level of \circ being judged as acceptable.

(14) Gel fraction (%): xylene immersion method, 120°C, 20 hours

- 10 (15) Tensile strength (MPa) and elongation (%): According to JIS K 6760

(16) Heat deformation (%): According to JIS K6723

- A polyolefinic base polymer and the carrier polymers A and B prepared above were blended in the proportions shown in Tables 3 and 4, the blend was extruded with an extruder into a tape and the tape was immersed in warm water, to effect crosslinking. The extruded tape was evaluated for its gel fraction, tensile strength, elongation and heat distortion.
- 15

Table 1

Compounding ingredient	A1	A2	A3	A4	A5	A6	A7
EEA	95	100		95	95	95	95
SEPS	5			5	5	5	5
L-LDPE			100				
VTMOS	45	40	40	100	2	45	45
DCP	2.16	1.92	1.92	4.8	0.5	0.2	20
Silane impregnability	○	○	×	×	○	○	○

Table 2

Compounding ingredient	B1	B2	B3	B4
LDPE	100		100	100
PP		100		
DBTDL	5	5	0.5	25
Antioxidant	16	16	16	16
Lubricant	5	5	5	5

Table 3

	Example			Comparative Example			
	1	2	3	1	2	3	4
<Compounding ingredient>							
LDPE	93.5			93.5			93.5
L-LDPE		93			93.5		
MDPE			92.5			93.5	
A1	5						5
A2		5.5	5.5				
A5				5			
A6					5		
A7						5	
B1	1.5		2		1.5	1.5	
B2		1.5		1.6			
B3							1.5
<Evaluation item>							
Extruded tape appearance	○	○	○	×	○	×	○
Gel fraction (%)	75	74	76	5	10	80	25
Tensile strength (M Pa)	19	21	23	-	18	-	18
Elongation (%)	450	500	470	-	490	-	460
Heat deformation (%)	12	8	7	-	-	-	-
Overall evaluation	○	○	○	×	×	×	×

Note: Symbol - means that determination was impossible.

Table 4

	Comparative Example		
	5	6	7
<Compounding ingredient>			
LDPE		97.5	
L-LDPE	93.5		
MDPE			81.5
A1		1	
A2	5		17
B1		1.5	
B2			1.5
B4	1.5		
<Evaluation item>			
Extruded tape appearance	x	o	x
Gel fraction (%)	78	20	82
Tensile strength (MPa)	-	18	-
Elongation (%)	-	460	-
Heat deformation (%)	-	-	-
Overall evaluation	x	x	x

Note: Symbol - means that determination was impossible.

As is apparent from Tables 3 and 4, the materials shown in Examples 1, 2 and 3 show good extrusion processability and quite excellent crosslinking property, mechanical property and heat
5 resistance.

In contrast, the materials shown in Comparative Examples are all poorly balanced among extrusion processability, crosslinking property, mechanical property and heat resistance.

10 Industrial Applicability

The process of the present invention is a very useful silane crosslinking process which can provide silane-crosslinked polyolefins excellent in extrusion processability as well as in crosslinking
15 property, mechanical property and heat resistance.

CLAIMS

1. A process for producing a silane-crosslinked polyolefin which is characterized by melt-blending (i) polyolefinic base polymer, (ii) a carrier polymer A
5 containing an organic unsaturated silane represented by the formula $RR'SiY_2$, wherein R is a monovalent olefinic unsaturated hydrocarbon group, Y is a hydrolyzable organic group, and R' is a monovalent hydrocarbon group other than aliphatic unsaturated hydrocarbon groups or
10 is the same group as Y, and a free radical generating agent and containing substantially no water and (iii) a carrier polymer B containing a silanol condensation catalyst and an antioxidant, at a temperature higher than the crystal melting point of the base polymer to
15 effect reaction, and then contacting the resulting reaction product with water to effect crosslinking.
2. The process for producing a silane-crosslinked polyolefin according to claim 1 wherein the base polymer is selected from the group consisting of
20 polyethylene, polypropylene, copolymer of ethylene and an α -olefin, ethylene-ethyl acrylate copolymer (EEA), ethylene-methyl methacrylate copolymer (EMMA), ethylene-vinyl acetate copolymer (EVA), chlorinated polyethylene (CPE), and the mixtures thereof.
- 25 3. The process for producing a silane-crosslinked polyolefin according to claim 1 wherein the carrier polymer A is selected from the group consisting

of ethylene-ethyl acrylate copolymer (EEA), ethylene-methyl methacrylate copolymer (EMMA), a hydrogenated block copolymer obtained by hydrogenating a block copolymer consisting essentially of a polymer block
5 comprising at least one vinylaromatic compound as the principal monomer unit and a polymer block comprising at least one conjugated diene compound as the principal monomer unit, and the mixtures thereof.

4. The process for producing a silane-
10 crosslinked polyolefin according to claim 1 wherein the carrier polymer B is selected from the group consisting of polyethylene, polypropylene, copolymer of ethylene and an α -olefin, and the mixtures thereof.

5. The process for producing a silane-
15 crosslinked polyolefin according to claim 1 wherein the total amount of the carrier polymer A and the carrier polymer B is 3-15% by weight.

6. The process for producing a silane-
20 crosslinked polyolefin according to claim 1 wherein the amount of the organic unsaturated silane added in the carrier polymer A is 0.1-5% by weight based on the total weight of the silane-crosslinked polyolefin.

7. The process for producing a silane-
25 crosslinked polyolefin according to claim 1 wherein the amount of the free radical generating agent added in the carrier polymer A is 0.01-0.5% by weight based on the total weight of the silane-crosslinked polyolefin.

8. The process for producing a silane-

crosslinked polyolefin according to claim 1 wherein the amount of the silanol condensaiton catalyst added in the carrier polymer B is 0.01-0.2% by weight based on the total weight of silane-crosslinked polyolefin.

INTERNATIONAL SEARCH REPORT

Int: onal Application No
PCT/JP 96/02712

A. CLASSIFICATION OF SUBJECT MATTER IPC 6 C08L43/04 C08J3/22		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC 6 C08L C08J		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practical, search terms used)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DE,A,31 50 808 (DIPOX KURT SCHULZE KG) 30 June 1983 see complete document ---	1-8
A	EP,A,0 426 073 (UNION CARBIDE CHEM PLASTIC) 8 May 1991 see page 3, line 10 - page 5, line 22; claims 1,22 ---	1
A	EP,A,0 416 851 (DU PONT CANADA) 13 March 1991 see page 3, line 5 - page 6, line 8 ---	1
A	DE,A,26 17 108 (KABEL METALLWERKE GHM) 27 October 1977 see claims 1,7 -----	1
<input type="checkbox"/> Further documents are listed in the continuation of box C. <input checked="" type="checkbox"/> Patent family members are listed in annex.		
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Date of the actual completion of the international search <div style="text-align: center; font-size: 1.2em;">15 January 1997</div>		Date of mailing of the international search report <div style="text-align: center; font-size: 1.2em;">27. 01. 97</div>
Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016		Authorized officer <div style="text-align: center; font-size: 1.2em;">Goovaerts, R</div>

INTERNATIONAL SEARCH REPORT

Information on patent family members

Inte: onal Application No

PCT/JP 96/02712

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
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